SEAI

CCE(P)-2015

KTM-06-XV

DO NOT OPEN THIS TEST BOOKLET UNTIL YOU ARE ASKED TO DO SO

Subject Code: 0 6

Test Booklet No.: 00294

TEST BOOKLET

CHEMICAL ENGINEERING

Time Allowed: 2 (Two) Hours

Full Marks: 200

INSTRUCTIONS

- 1. The name of the Subject, Roll Number as mentioned in the Admission Certificate, Test Booklet No. and Subject Code shall be written legibly and correctly in the space provided on the Answer Sheet with black ball pen.
- 2. Space provided for Series in the Answer Sheet is not applicable for Optional Subject. So the space shall be left blank.
- 3. All questions carry equal marks. Your total marks will depend only on the number of correct responses marked by you in the Answer Sheet.
- 4. No candidate shall be admitted to the Examination Hall/Room 20 minutes after commencement of distribution of the paper. The Supervisor of the Examination Hall/Room will be the time-keeper and his/her decision in this regard is final.
- 5. No candidate shall leave the Examination Hall/Room without prior permission of the Supervisor/
 Invigilator. No candidate shall be permitted to hand over his/her Answer Sheet and leave the
 Examination Hall/Room before expiry of the full time allotted for each paper.
- 6. No Mobile Phone, Pager, etc., are allowed to be carried inside the Examination Hall/Room by the candidates. Any Mobile Phone, Pager, etc., found in possession of the candidate inside the Examination Hall/Room, even if on off mode, shall be liable for confiscation.
- 7. No candidate shall have in his/her possession inside the Examination Hall/Room any book, notebook or loose paper, except his/her Admission Certificate and other connected paper permitted by the Commission.
- 8. Complete silence must be observed in the Examination Hall/Room. No candidate shall copy from the paper of any other candidate, or permit his/her own paper to be copied, or give, or attempt to give, or obtain, or attempt to obtain irregular assistance of any kind.
- 9. After you have completed filling in all your responses on the Answer Sheet and the Examination has concluded, you should hand over to the Invigilator only the Answer Sheet. You are permitted to take away with you the Test Booklet.
- 10. Violation of any of the above Rules will render the candidate liable to expulsion from the Examination Hall/Room and disqualification from the Examination, and according to the nature and gravity of his/her offence, he/she may be debarred from future Examinations and Interviews conducted by the Commission for appointment to Government Service.
- 11. Smoking inside the Examination Hall/Room is strictly prohibited.
- 12. This Test Booklet contains one sheet (two pages) for Rough Work at the end.

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[No. of Questions: 100]

CCE(P) - 2015 CHEMICAL ENGINEERING

(A) Sublimation (B) Sulphonation (C) Pyrolysis (D) Hydrolysis (E) Which of the following is not equivalent to standard atmospheric pressure? (A) 1.01325 Pa (B) 1.0135 bar (C) 760 mm Hg (D) 1.01325 × 10 ⁵ N/m ² (C) flow area to wetted perimeter (D) square root of flow area wetted perimeter (A) one molecule of the substance (B) laminar flow through an operation of the above (C) 6.023 × 10 ²³ molecules (D) None of the above (E) laminar flow through an operation of the above (C) steady flow (D) None of the above (E) laminar flow through an operation of the above (C) steady flow (D) None of the above (D) None of the above (E) laminar flow through an operation of the above (C) steady flow (D) None of the above	(A) Sublimation	(A) varies parabolically across the
(B) Sulphonation (C) Pyrolysis (D) Hydrolysis (D) Hydrolysis (E) is zero at the centre and varie linearly with radius (D) is zero at the wall an increases linearly to the centre viscosity of a liquid (A) increases (B) decreases (C) remains constant (D) None of the above (E) 1-01325 Pa (C) 760 mm Hg (D) 1-01325 \times 10 ⁵ N/m ² (C) flow area to wetted perimeter (D) square root of flow area wetted perimeter		Cross-section
(C) Pyrolysis (C) is zero at the centre and varie linearly with radius (D) is zero at the wall an increases linearly to the centre viscosity of a liquid (A) increases (A) 1.01325 Pa (B) 1.0135 bar (C) 760 mm Hg (D) 1.01325 × 10 ⁵ N/m ² (C) 1.01325 × 10 ⁵ N/m ² (D) 1.01325 × 10 ⁵ N/m ² (E) 1.01325 × 10 ⁵ N/m ² (C) 1.01325 × 10 ⁵ N/m ² (C) 1.01325 × 10 ⁵ N/m ² (D) 1.01325 × 10 ⁵ N/m ² (E) 1.01325 × 10 ⁵ N/m ² (C) 1.01325 × 10 ⁵ N/m ² (C) 1.01325 × 10 ⁵ N/m ² (D) 1.01325 × 10 ⁵ N/m ² (E) 1.01325 × 10 ⁵ N/m ² (C) 1.01325 × 10 ⁵ N/m ² (D) 1.01325 × 10 ⁵ N/m ² (E) 1.01325 × 10 ⁵ N/m ² (C) 1.01325 × 10 ⁵ N/m ² (D) 1.01325 × 10 ⁵ N/m ² (E) 1.01325 × 10 ⁵ N/m ² (C) 1.01325 × 10 ⁵ N/m ² (D) 1.01325 × 10 ⁵ N/m ² (E) 1.01325 × 10 ⁵ N/m ² (D) 1.01325 × 10 ⁵ N/m ² (E) 1.01325 × 10 ⁵ N/m ² (D) 1.01325 × 10 ⁵ N/m ² (E) 1.01325 × 10 ⁵ N/m ² (D) 1.01325 × 10 ⁵ N/m ² (E) 1.01325 × 10 ⁵ N/m ² (D) 1.01325 × 10 ⁵ N/m ² (E) 1.01325 × 10 ⁵ N/m ² (D) 1.01		
(C) Pyrolysis (D) Hydrolysis (D) is zero at the wall an increases linearly to the centre of the following is not equivalent to standard atmospheric pressure? (A) 1.01325 Pa (B) 1.0135 bar (C) 760 mm Hg (C) 760 mm Hg (D) 1.01325 \times 10 ⁵ N/m ² (E) flow area to wetted perimeter to flow area wetted perimeter (C) flow area to square of wetter perimeter (D) square root of flow area wetted perimeter (D) square root of flow area wetted perimeter (D) square root of flow area wetted perimeter (E) square root of flow area wetted perimeter (D) square root of flow area wetted perimeter (E) square root of flow area wetted perimeter (D) square root of flow area wetted perimeter (E) square flow (B) laminar flow through an oper channel (C) steady flow	(B) Sulphonation	PEZUU
(D) Hydrolysis 5. With increase in the temperature viscosity of a liquid (A) increases (A) 1.01325 Pa (B) 1.01325 Pa (C) 760 mm Hg (D) 1.01325 × 10 ⁵ N/m ² (E) flow area to wetted perimeter (D) square root of flow area wetted perimeter (E) square root of flow area wetted perimeter (D) square root of flow area wetted perimeter (E) square root of flow area wetted perimeter (D) square root of flow area wetted perimeter (E) square root of flow area wetted perimeter	(C) Pyrolysis	
5. With increase in the temperature viscosity of a liquid not equivalent to standard atmospheric pressure? (A) 1.01325 Pa (B) 1.0135 bar (C) 760 mm Hg (B) 1.01325 × 10 ⁵ N/m ² (C) 1.01325 × 10 ⁵ N/m ² (D) 1.01325 × 10 ⁵ N/m ² (E) 1.01325 × 10 ⁵ N/m ² (C) 1.01325 × 10 ⁵ N/m ² (D) 1.01325 × 10 ⁵ N/m ² (E) 1.01325 × 10 ⁵ N/m ² (C) 1.01325 × 10 ⁵ N/m ² (D) 1.01325 × 10 ⁵ N/m ² (E) 1.01325 × 10 ⁵ N/m ² (C) 1.01325 × 10 ⁵ N/m ² (D) 1.01325 × 10 ⁵ N/m ² (E) 1.01325 ×	BOOKLET	(D) is zero at the wall an
Which of the following is not equivalent to standard atmospheric pressure? (A) 1.01325 Pa (B) decreases (C) remains constant (D) None of the above (E) 1.0135 bar (C) 760 mm Hg (D) 1.01325 \times 10 ⁵ N/m ² (C) flow area to wetted perimeter (D) square root of flow area wetted perimeter (E) square root of flow area wetted perimeter (D) square root of flow area wetted perimeter (E) square root of flow area wetted perimeter	(D) Hydrolysis	increases linearly to the centre
Which of the following is not equivalent to standard atmospheric pressure? (A) 1.01325 Pa (B) decreases (C) remains constant (D) None of the above (E) 1.0135 bar (C) 760 mm Hg (D) 1.01325 × 10 ⁵ N/m ² (E) flow area to wetted perimeter (D) square root of flow area wetted perimeter (E) square flow area wetted perimeter (E) steady flow	Full Marks :	
atmospheric pressure? (A) 1.01325 Pa (B) decreases (C) remains constant (D) None of the above (E) 1.0135 bar (C) 760 mm Hg (D) 1.01325 × 10 ⁵ N/m ² (E) flow area to wetted perimeter (D) 1.01325 × 10 ⁵ N/m ² (C) flow area to square of wetter perimeter (D) square root of flow area wetted perimeter (D) square root of flow area wetted perimeter (E) square root of flow area wetted perimeter	. Which of the following is	
(A) 1.01325 Pa (C) remains constant (D) None of the above (E) 1.0135 bar (C) 760 mm Hg (E) 1.01325×10^5 N/m ² (D) 1.01325×10^5 N/m ² (E) flow area to wetted perimeter (C) flow area to square of wetter perimeter (D) square root of flow area wetted perimeter (D) square flow area wetted perimeter (D) square flow area wetted perimeter (E) square flow area wetted perimeter (D) square flow area wetted perimeter	consiste to standard	(A) increases
(A) 1.01325 Pa (C) remains constant (D) None of the above (E) 1.0135 bar (C) 760 mm Hg (A) wetted perimeter to flow area (B) flow area to wetted perimeter (C) flow area to square of wetter perimeter (D) square root of flow area wetted perimeter (D) square root of flow area wetted perimeter (D) one molecule of the substance (E) $f(x) = \frac{16}{N_{Re}}$ is valid for (E) one atom of the substance (A) turbulent flow (B) laminar flow through an operimeter (C) steady flow	atmospheric pressure?	(D) accidates
(D) None of the above (B) 1.0135 bar 6. Hydraulic radius is the ratio of (C) 760 mm Hg (A) wetted perimeter to flow area (B) flow area to wetted perimeter (C) flow area to square of wetter perimeter (D) square root of flow area wetted perimeter (A) one molecule of the substance (B) flow area to square of wetter perimeter (C) flow area to square of wetter perimeter (D) square root of flow area wetted perimeter (E) square root of flow area wetted perimeter (E) square root of flow area wetted perimeter (E) laminar flow through an operation of the substance (C) steady flow	(A) 1.01325 Pa	(C) remains constant
(B) 1.0135 bar 6. Hydraulic radius is the ratio of (C) 760 mm Hg (A) wetted perimeter to flow area (B) flow area to wetted perimeter (C) flow area to square of wetter perimeter (D) square root of flow area wetted perimeter (A) one molecule of the substance (B) one atom of the substance (C) $f = \frac{16}{N_{Re}}$ is valid for (D) square root of flow area wetted perimeter (E) square root of flow area wetted perimeter	total marks will depend only on the number of equ	(D) None of the above
(C) 760 mm Hg (A) wetted perimeter to flow area (B) flow area to wetted perimeter (B) flow area to wetted perimeter (C) flow area to square of wetter perimeter (D) square root of flow area wetted perimeter (A) one molecule of the substance (B) one atom of the substance (C) $f = \frac{16}{N_{Re}}$ is valid for (D) square root of flow area wetted perimeter (E) square root of flow area wetted perimeter	(B) 1.0135 bar	5. Hydraulic radius is the ratio of
(D) $1.01325 \times 10^5 \text{ N/m}^2$ (C) flow area to square of wetter perimeter (D) square root of flow area wetted perimeter (A) one molecule of the substance (B) one atom of the substance (C) $f = \frac{16}{N_{\text{Re}}}$ is valid for (D) square root of flow area wetted perimeter (E) square root of flow area wetted perimeter	(C) 760 mm Hg	(A) wetted perimeter to flow area
(C) flow area to square of wetter perimeter (D) square root of flow area wetted perimeter (E) square root of flow area wetted perimeter (E) one molecule of the substance (E) one atom of the substance (E) one atom of the substance (C) 6.023×10^{23} molecules (C) steady flow		(B) How area to wetten permitte
(D) square root of flow area wetted perimeter (A) one molecule of the substance (B) one atom of the substance (C) 6.023×10^{23} molecules (D) square root of flow area wetted perimeter (A) turbulent flow (B) laminar flow through an oper channel (C) steady flow	(D) 1.01325 × 10° N/m²	(C) flow area to square of wetter perimeter
(B) one atom of the substance (B) laminar flow through an oper channel (C) 6.023 × 10 ²³ molecules (C) steady flow	3. One mole of a compound contains	(D) square root of flow area wetted perimeter
(B) one atom of the substance (A) turbulent flow (B) laminar flow through an operation of the substance (C) 6.023×10^{23} molecules (C) steady flow	(A) one molecule of the substance	7. $f = \frac{16}{N_{\text{Re}}}$ is valid for
(C) 6.023×10^{23} molecules (C) steady flow	(B) one atom of the substance	(A) turbulent flow
converse instruction in a minimum (C) steady now	(C) 6.023×10^{23} molecules	(B) laminar flow through an open
(D) 22.4×10^3 molecules (D) None of the above		(C) steady now
	(D) 22.4×10^3 molecules	(D) None of the above

8.	Limiting	rea	ctar	nt f	or	a	che	mical
	equation	is	the	rea	cta	nt.	that	is

- (A) present in excess of the stoichiometric amount
- (B) present exactly as per stoichiometric amount
- (C) present in the smallest stoichiometric amount
 - (D) not having any relation with stoichiometry
 - 9. When chemical reaction occurs in a process, then usually the number of independent material balances and the number of atomic species
- (A) is different
 - (B) is equal
 - (C) may or may not matter
 - (D) is irrelevant
 - The amount of air or oxygen required for bringing a process into complete combustion is called
 - (A) theoretical air or oxygen
 - (B) excess air or oxygen
 - (C) practical requirement
 - (D) None of the above

- 11. With rise in temperature, the heat capacity of a substance
 - (A) increases
 - (B) decreases
 - (C) remains unchanged
 - (D) either (A) or (B) depends on the substance
- 12. A stream bled off to remove an accumulation of inerts or unwanted material in the process is called
 - (A) bypass stream
 - (B) purge stream
- (C) recycle stream
 - (D) process stream
 - 13. In general, a gas that exists below its critical temperature is called
 - (A) a critical gas
 - (B) a condensate
 - (C) a vapour
 - (D) a smoke to smok (C)

- 14. Gibbs' phase rule is given by
 - (A) F + C = P + 2
 - (B) F C = P 2
 - (C) F = C + P + 2
 - (D) F = C P + 2
- 15. The forces acting on a particle settling in a fluid are
 - (A) gravitational and buoyant forces

(B) decreases

- (B) centrifugal and drag forces
- (C) gravitational or centrifugal, buoyant and drag forces
- (D) external, drag and viscous forces
- 16. If more than two branches of pipes are to be connected at the same point, then which of the following is used?
 - (A) Elbow as facilities a (A)
 - (B) Union Sansbarra a (B)
 - (C) Tee
 - (D) None of the above

- 17. Pressure drop in a packed bed for laminar flow is given by
 - (A) Kozeny-Carman equation
 - (B) Blake-Plummer equation
 - (C) Leva's equation
 - (D) None of them
- 18. How many atoms are there per unit cell in a body-centred cubic lattice system?
 - (A) 2
 - (B) 3
 - (C) 4
 - (D) 6
 - 19. A material is able to retain the deformation permanently by virtue of its
 - (A) elasticity
 - (B) plasticity
 - (C) ductility
- (D) malleability
 - 20. Which one is a non-magnetic material?
 - (A) Cobalt
 - (B) Nickel
 - (C) Zinc
 - (D) None of the above

21. Which metal is protected by the layer	25. Teflon is corroded by
of its own oxide?	(A) hydrochloric acid (10%)
(A) . Iron singlem (A)	(B) hydrochloric acid (95%)
(B) Silver	(C) sulphuric acid
(C) Calcium	(D) None of the above
(D) Aluminium (D)	(D) All of the above
22. The viscosity of atmospheric air may be about centipoise.	26. Vena contracta formed during flow of a liquid through an orifice meter has
(A) 1.5	(A) minimum liquid cross-section
(B) 15	(B) more diameter compared to orifice diameter
(C) 150 (O)	(C) minimum velocity of fluid stream
(D) 0.015 x55m tatevia (C)	(D) None of the above
23. The quantity of heat required to evaporate 1 kg of a saturated liquid is called (A) specific heat	27. "The total volume occupied by a gaseous mixture is equal to the sum of the pure component volume." This is
(B) 1 kcal	(A) Dalton's law ModA (O)
(C) 1 calorie atidgaro (5)	(B) Amagat's law down (C)
(D) latent heat	(C) Gay-Lussac law (D) Avogadro's law
24. The softest material in Mohs scale is	woll birdless bus well sind took in it
(A) talc	28. Plunger pumps are used for
(B) an clastomer	(A) higher pressure
(C) an alloy of tin and lead	(B) slurries
(C) rubber	(C) viscous mass
(D) iron contact the contact that the contact the cont	(D) None of the above

29. The total energy at of energy.			per classification of materials, er glass' may be classified under
(A) potential an	d kinetic	(A)	metals And A
(B) pressure	rudqlue (3)	(B)	ceramics (a)
(C) internal	(D) None o	(C)	composites
(D) All of the al	oove	(D)	polymers
30. An example of no may be	on-Newtonian fluid	divi	volume of atoms in a unit cell ded by total volume of the unit
(A) ideal gas		cell	is known as
(B) thin liquid		(A)	atomic number
	chain hydrocarbon	(B)	atomic packing factor
(D) None of the		(C)	coordination number
f the above		(D)	crystal index
31. Which is not	a dimensionless		
parameter?	27. The total v	35. Whic	h among the following is not a
(A) Reynolds nu			oon polymorph?
the pure component			is called
(B) Froude num		(A)	Diamond
(C) Atomic num	ber (A)	(B)	Pearl
(D) Mach numb	er am A (B)	(C)	Graphite (D)
32. The concept of bo	oundary layer that	(D)	Fullerene tead metal (G)
ideal-fluid flow a	and real-fluid flow	36. Pers	
was given by	28. Plunger pun		24. The softest material in Mol
		(A)	Acrylic sheet
(A) Reynolds	(A) higher	(B)	an elastomer
(B) Prandtl		(C)	an alloy of tin and lead
(C) Mach	(C) viscous	(D)	an aluminium foil clad with
(D) Hagen	O SHOW (CD)		bakelite

37. Atmospheric corrosion of metals results from their	41. Which of the following produces maximum pressure difference for transportation of gases?
(A) slow oxidation (B) fast oxidation	(A) Vacuum pump
(C) fast hydration (D) slow hydration	(C) Fan (D) Compressor
50. Descration of water in its treatment	(D) pulverised coal in boilers
38. Boiler feed-water is treated to prevent (A) scaling and corrosion (B) foaming and priming (C) Both (A) and (B) (D) Neither (A) nor (B)	 42. The number of gram equivalent of solute dissolved in one litre of solution is called its (A) normality (B) molarity (C) molality (D) None of the above
coagulant for removal of suspended	(b) Note of the above
39. Which of the following is not a renewable energy?	43. Rubber-lined pumps can be used to pump
(A) Solar energy	(A) caustic soda
2012 (0)	(B) chlorinated brine
(B) Wind energy	(C) hypochloric acid
(C) Nuclear energy (D) Geothermal energy	(D) All of the above
(A) yields acid of higher	44. In flue gas analysis by Orsat's
40. Glass is corroded by	apparatus, carbon monoxide is absorbed by
(A) fluorine (dry or wet)	(A) cuprous chloride
(B) H ₂ SO ₄	(B) potassium hydroxide
(C) H ₃ PO ₄ aratosdo at (O)	(C) alkaline pyrogallol solution
(D) None of the above	(D) None of the above

- 45. Emission of dense white smoke out of the chimney of a thermal power plant is an indication of the use of
 - (A) less air for combustion
 - (B) correct amount of air for combustion
 - (C) too much air for combustion
 - (D) pulverised coal in boilers

42. The mumber of gram equivalent of

- 46. A coal that softens and fuses on heating is
 - (A) classified
 - (B) carbonised
 - (C) caking
 - (D) non-caking
- 47. The calorific value is highest for
 - (A) producer gas
 - (B) water gas
 - (C) coke-oven gas
 - (D) blast-furnace gas
- 48. Addition of a non-volatile solute to a solvent produces a ____ in its solvent.
 - (A) freezing point elevation
 - (B) boiling point depression
 - (C) vapour pressure lowering
 - (D) None of the above

- **49.** Recycling in a chemical process facilitates
 - (A) increased yield
 - (B) enrichment of product
 - (C) heat conservation
 - (D) All of the above
- **50.** Deaeration of water in its treatment is necessary as it
 - (A) minimises its turbidity
 - (B) helps in controlling its taste and odour
 - (C) minimises its corrosiveness
 - (D) None of the above
- **51.** The most widely and commonly used coagulant for removal of suspended impurities in water is
 - (A) bleaching powder
 - (B) slaked lime
 - (C) alum yarana barW (E)
 - (D) copper sulphate
- 52. Contact process
 - (A) yields acid of higher concentration than chamber process
 - (B) yields acid of lower concentration than chamber process
 - (C) is obsolete
 - (D) eliminates absorber

53. Bromine is used in the preparation of	57. Turbulent flow generally occurs in cases involving
(A) fire extinguishing compounds	(A) highly viscous fluid
(B) fireproofing agents	(B) very narrow passages
(C) dyes and antiknock compounds	(C) very slow motion
(D) All of the above	(D) None of the above
54. The hottest part of a flame lies in its	58. Cement mainly contains
(A) non-luminous zone	(A) CaO, SiO ₂ , Al ₂ O ₃
(B) luminous zone	(B) MgO, SiO ₂ , K ₂ O
(C) yellow zone	(C) Al ₂ O ₃ , MgO, Fe ₂ O ₃
(D) zone of unburnt gases	(D) CaO, MgO, K ₂ O
55. When coal is heated in absence of air, it is called	59. Absorption of SO ₃ in 97% H ₂ SO ₄ is
(A) deoxidation	(A) exothermic
(B) gasification	(B) endothermic
(C) coalification	(C) not possible
(D) carbonisation	(D) None of the above
56. Which is the most commonly used molten metal for cooling of nuclear reactor?	60. Which form of energy source has the highest potential in Assam compared to all States of India?
(A) Calcium (A)	(A) Wind power XBW (A)
(B) Sodium (B)	(B) Solar power diagon (B)
(C) Mercury	(C) Hydropower (C)
(D) Zinc stoneut shem (C)	(D) Biomass power (D)

61.	Find	the odd one out.	65. Was	hing soda is u ai salmona E
	(A)	Digboi Refinery		Na ₂ CO ₃
	20 20	Numaligarh Refinery		Na ₂ CO ₃ ·H ₂ O
		Guwahati Refinery		Na ₂ CO ₃ · 10H ₂ O
	(D)	Bongaigaon Refinery		NaHCO ₃
62.		etrating index and softening		main purpose of galvanizing iron ets is to
	(A)	gasoline de de de la	(A)	harden the surface
	(B)	kerosine OR OM (F)	(B)	increase the glossiness and luster
	(C)	bitumen M (EOSIA (O)		nuster onas wallsy (2)
	(D)	gas oil OsM OsO (a)	(C)	prevent the action of water
			(D)	prevent the action of oxygen
63.	Merc	cury is transported in metal		55. When coal is heated in a
		tainers made of	67. Amm	nonia synthesis reaction is
	(A)	aluminium (8)	(A)	exothermic
	(B)	lead . (C) not possible	(B)	endothermic
	(C)	iron avoda edi to enoli (C)	(C)	autocatalytic
	(D)	nickel	(D)	None of the above
		60. Which form of energy s	beau vinor	
		crude means the crude oil	68. A ma	aterial is called ductile if it can
	(A)	wax reword brief (A)	(A)	drawn into wire
	(B)	paraffins and asphalts	(B)	hammered into a thin sheet
	(C)	nitrogen compounds	(C)	fractured without deformation
	(D)	sulphur compounds	(D)	made lustrous by heating it

69. Which is an amorphous material?	73. The most commonly used joint in underground pipeline is the
(A) Glass	(A) sleeve joint Market (A)
(B) Mica godino IstoT (A)	(B) coupling
(C) Brass Roding box (A)	(C) flange (consequence)
(D) Cast iron (D)	(D) expansion joint (2)
70. Which of the following is the easiest to crack?	74. Flow measurement in open channels is done by
(A) Paraffin	(A) venturi meter
(B) Olefin moitalitied (A)	(B) orifice meter vinsionis
(C) Naphthene	(C) weir (A)
(D) Aromatic nongroud (D)	(D) rotameter (S)
71. Laminar flow of a Newtonian fluid ceases to exist, when Reynolds	75. The catalyst used in catalytic reforming is
number exceeds bos don't all a company (A) 4000	(A) platinum on alumina (B) nickel
(B) 2100	(C) iron a Isrophevnos (A)
(C) 1500	(D) aluminium chloride
(D) Phosphoric acid	76. When you completely burn the coal
72. Head developed by a centrifugal pump depends on its	sample in a muffled furnace at 700 °C-750 °C and weigh the residue you get
(A) speed Sq. (A)	(A) volatile matter
(B) impeller diameter (8)	(B) fixed carbon
(C) Both (A) and (B) (3)	(C) ash content pented (C)
(D) None of the above	(D) mineral matter (C)

- 77. Pick the correct statement from the following:
 - (A) Synthesis gas is a mixture of CO, CO₂ and H₂.
 - (B) Synthesis gas is a mixture of CO₂ and H₂.
 - (C) Synthesis gas is a mixture of O₂ and H₂.
- (D) None of the above
- 78. The specific type of uranium used to produce nuclear energy most efficiently is
 - (A) U-233
 - (B) U-234
 - (C) U-235
- (D) U-236
- 79. A hydel project that captures the kinetic energy of water without large reservoir or even dam is called
 - (A) conventional project
 - (B) run-of-the-river project
 - (C) micro-hydro project
 - (D) conduit project
- **80.** A solution with reasonably permanent pH is called a/an
 - (A) non-ideal solution
 - (B) colloidal solution
 - (C) buffer solution
 - (D) ideal solution

- 81. Which of the following cannot be determined by ultimate analysis of coal?
 - (A) Total carbon
 - (B) Fixed carbon
 - (C) Hydrogen
 - (D) Oxygen
- **82.** Which among the following is a unit process?
 - (A) Distillation
 - (B) Hydrogenation of oils
 - (C) Absorption
 - (D) Humidification
- 83. Which acid is used to the greatest extent for metal pickling?
 - (A) Sulphuric acid
 - (B) Hydrochloric acid
 - (C) Nitric acid
 - (D) Phosphoric acid
- **84.** For an ideal gas, $C_p C_v = ?$
 - (A) R^2
 - (B) R/2 the relieque (E)
 - (C) R (8) one (A) rho8 (D)
 - (D) 2R da on to snow (Th

85. Kerosine should have	
(A) low smoke point	a good commercial sample of bleaching powder is
(B) high smoke point	(A) 15% to 17%
(C) high aromatic content	(B) 35% to 37%
(D) low paraffin content	(D) 69% to 71%
86. Priming is needed in a off. 80	90. Cement setting under water employs a/an process.
(A) reciprocating pump	(A) hydration
(B) gear pump	(B) decomposition
(C) centrifugal pump	(C) oxidation
(D) diaphragm pump	(D) reduction
87. The catalyst used in Fischer-Tropsch process is	91. Triple superphosphate is made by reacting phosphoric rock with
(A) nickel tines but in (A).	(A) phosphoric acid
(B) zinc oxide	(B) nitric acid
(C) alumina (D) thorium oxide	(C) sulphuric acid (D) hydrochloric acid
100. The most common method to	92. After feeding raw materials to the
88. Drag force on the float of a rotameter	clinkers, what is the correct order of reaction zones?
(A) ∞ Q (A)	(A) Calcining-Drying-Clinkering
(B) ∝√Q (a)	(B) Drying-Calcining-Clinkering
(C) ∝ Q ² mixeld masts (D)	(C) Clinkering-Drying-Calcining
(D) is constant algraes (C)	(D) Clinkering-Calcining-Drying

- 93. A turbine is used to convert
 - (A) kinetic energy to mechanical energy
 - (B) mechanical energy to hydraulic energy
 - (C) hydraulic energy to mechanical energy
 - (D) None of the above
- **94.** Junker's calorimeter is used to determine the calorific value of
 - (A) gaseous fuel
 - (B) liquid fuel
 - (C) pulverised coal
 - (D) solid fuel
- 95. The main constituent of natural gas is
 - (A) CH₄
 - (B) C₂H₂
 - (C) C₂H₄
 - (D) C₂H₆
- 96. Which of the following fraction of a crude will have maximum API (i.e., °API)?
 - (A) Diesel
 - (B) Gasoline (B)
 - (C) Atmospheric gas oil
 - (D) Vacuum gas oil

- **97.** DCDA process is the most recent process for the manufacture of
 - (A) HNO₃
 - (B) H₂SO₄
 - (C) HCl
 - (D) None of the above
- **98.** The widely used method for conditioning of boiler feed-water is
 - (A) cold lime process
 - (B) coagulation
 - (C) hot lime process
 - (D) sequestration
- 99. In a centrifugal pump, the liquid enters the pump
 - (A) at the centre
 - (B) at the bottom
 - (C) at the top
 - (D) from the sides
- 100. The most common method of producing hydrogen from hydrocarbon is known as
 - (A) steam reforming
 - (B) partial oxidation
 - (C) steam blowing
 - (D) complete oxidation